

## ANALYSIS OF URANIUM BY TRLIF, RIMS AND ICP-MS

Laser spectroscopy (Resonance Ionisation Mass Spectrometry-RIMS, Time Resolved Laser Induced Fluorescence-TRLIF, Time Resolved Laser Induced Chemiluminescence-TRLIC,) and Inductively Coupled Plasma Mass Spectrometry (ICP-MS) can be very efficient for elemental and isotope composition analysis of various samples, as well as for the determination of the molecular and valence forms of uranium (speciation analysis) [1-5]. A series of reference materials measurements with various isotope compositions ranging from depleted and natural to enriched uranium by RIMS have been previously reported by our collaboration [1,2]. For samples of depleted uranium the  $^{235}\text{U}/^{238}\text{U} < 0.003$  ratio was determined with  $<7\%$  precision ( $2\sigma$  errors) for the total uranium concentrations not exceeding  $\sim 80$  fg per sample [1].

Without mineralization, the limit of uranyl detection (LOD) by TRLIF in blood plasma has been determined  $0.1\text{ng/ml}$ . After mineralization, a lower LOD ranging  $0.008\text{ng/ml} - 0.01\text{ng/ml}$  has been evaluated. The limit of uranyl detection in urine in our TRLIF experiments was up to  $0.005\text{ng/ml}$ . Such LOD are sufficient to allow for studies the dynamics processes and behaviour of the of uranium in biologicals objects [3,4]. However, actinides in various valence states do not all exhibit luminescence properties and for such cases the TRLIC methods can be applied [2,5].

A high concentration of uranium we detected by ICP-MS in the bones of dinosaurs ( $122\text{mg/kg}$ ), South mammoth ( $220\text{mg/kg}$ ), prehistoric bear ( $24\text{mg/kg}$ ) and archanthropus ( $1.5\text{mg/kg}$ ) compared to surrounding soils ( $3.7\text{mg/kg} - 7.8\text{mg/kg}$ ) and standard bones ( $< 0.01\text{mg/kg}$ ) was established. The standart  $^{235}\text{U}/^{238}\text{U} = 0.007$  ratio was detected for all samples, but the  $^{234}\text{U}/^{238}\text{U}$  (detected  $1.6 \cdot 10^{-4} \div 5.8 \cdot 10^{-5}$ ) ratio differ from secular equilibrium value  $^{234}\text{U}/^{238}\text{U} (5.5 \cdot 10^{-5})$ .

### References

1. Strashnov I.M., et al. *Anal. Atom. Spectroscopy* 2019, **34**, 1630.
2. Strashnov I.M., et al. *J. of Radioanalytical and Nuclear Chemistry* 2019, **322**, 1437.
3. Izosimov I.N. *Procedia Chemistry* 2016, **21**, 473.
4. Izosimov I.N. *Environmental Radiochemical Analysis VI*, pp. 115-130. Royal Society of Chemistry Publishing, 2019.  
DOI: 10.1039/9781788017732-00115
5. Izosimov I.N. *J. of Radioanalytical and Nuclear Chemistry* 2015, **304**, 207.

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**Track Classification:** The V International Scientific Forum “Nuclear Science and Technologies”: Radiation ecology and methods of analysis (Section 3)